

Aryl C-CN Insertions

1-25-2025

Group Meeting

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Texas A&M University

Outline

Part I. Characterization of Aryl C-CN Insertion

Part II. Aryl C-CN Bond Activation in Synthesis

Historical Precedent

- DuPont (1971)

We find some parallel to the reactivity of $\text{Pt}[\text{P}(\text{C}_2\text{H}_5)_3]_4$ in the chemistry of $\text{Ni}[\text{P}(\text{C}_2\text{H}_5)_3]_4$.¹² This complex, an off-white crystalline compound, was formed almost quantitatively by reaction of bis(1,5-cyclooctadiene)nickel(0) with triethylphosphine. It dissolved in hydrocarbon solvents with ligand dissociation to give intensely purple solutions of $\text{Ni}[\text{P}(\text{C}_2\text{H}_5)_3]_3$. This tris complex reacted smoothly with chlorobenzene and with benzonitrile *at room temperature* to give the corresponding *trans*- $\text{C}_6\text{H}_5\text{NiX}[\text{P}(\text{C}_2\text{H}_5)_3]_2$ compounds.¹³ Work is in progress to examine other aspects of reactive d^{10} complexes of Ni, Pd, and Pt.

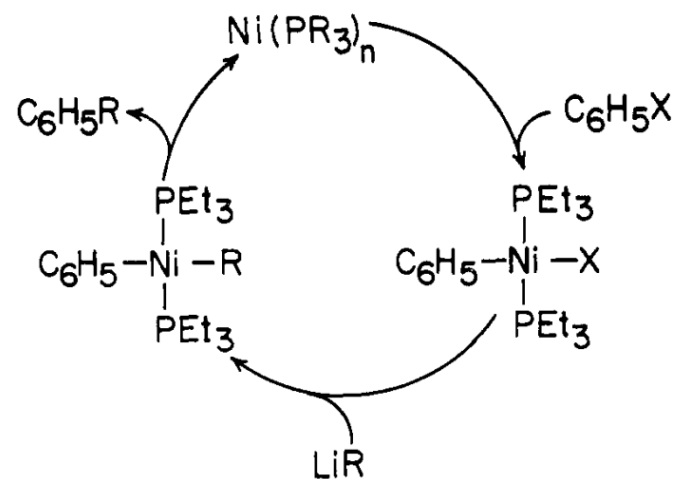
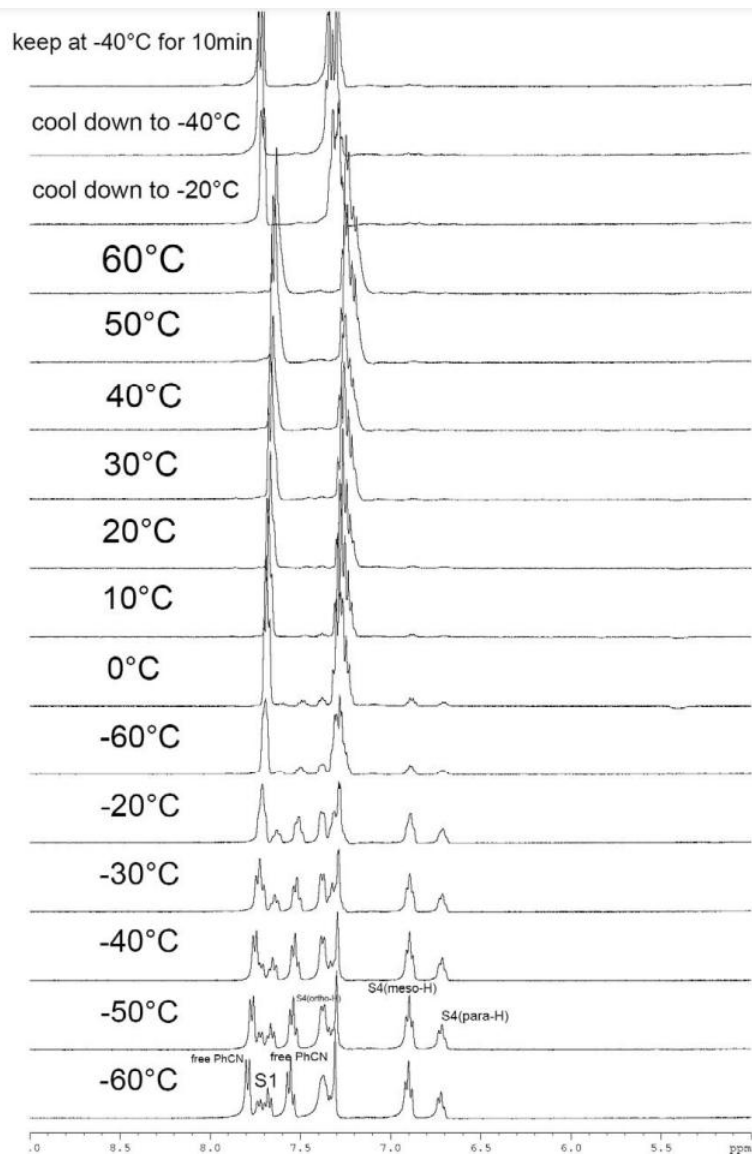
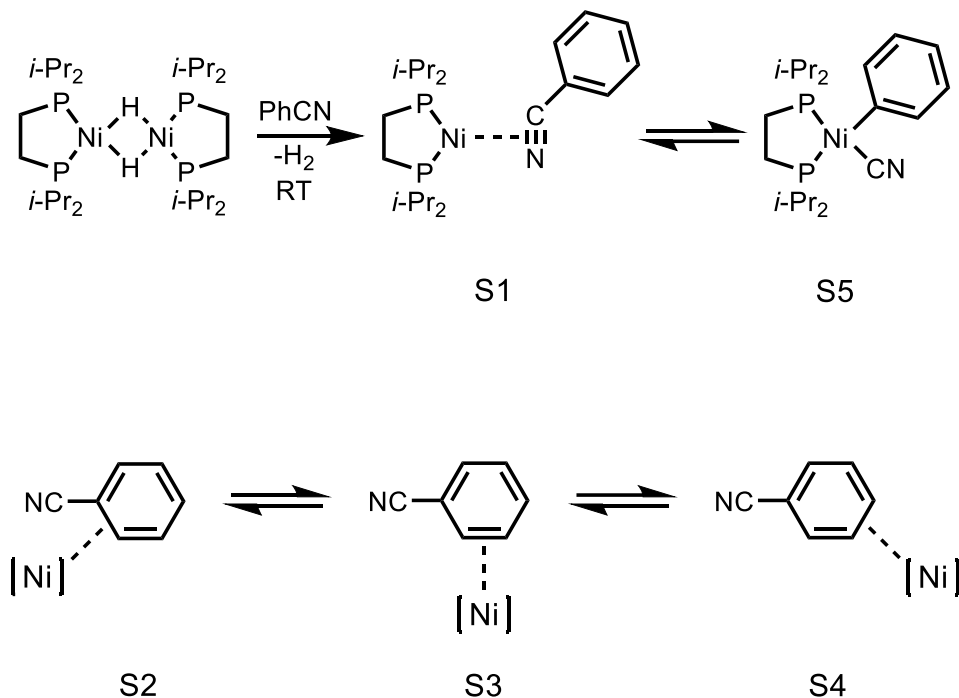


Figure 1. Cycle for alkylation, arylation, and cyanation of halo-benzenes catalyzed by zerovalent nickel complexes. R = alkyl, aryl, CN.

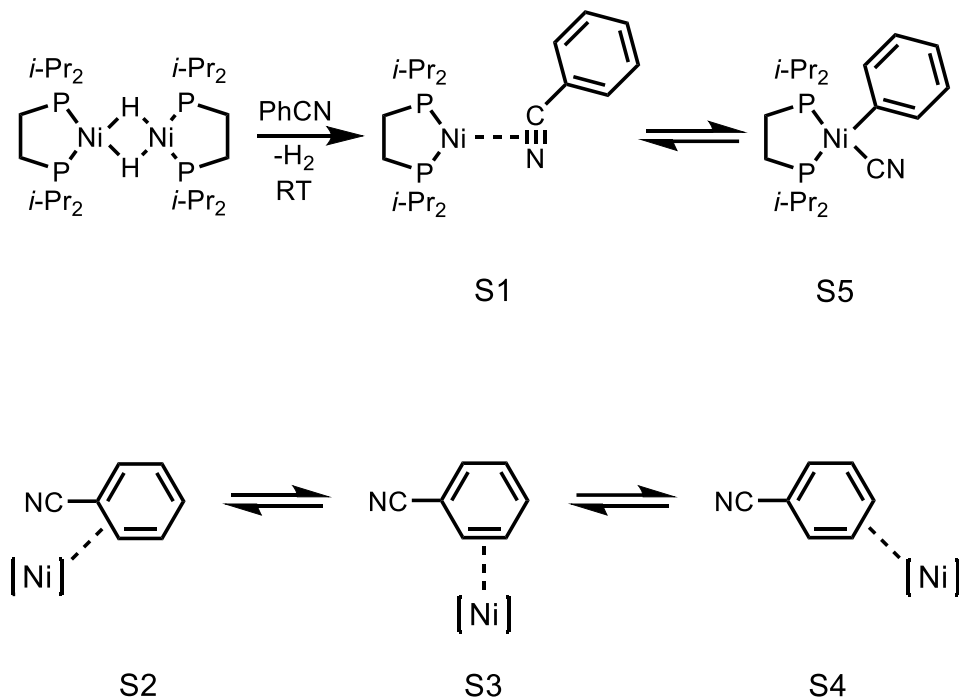
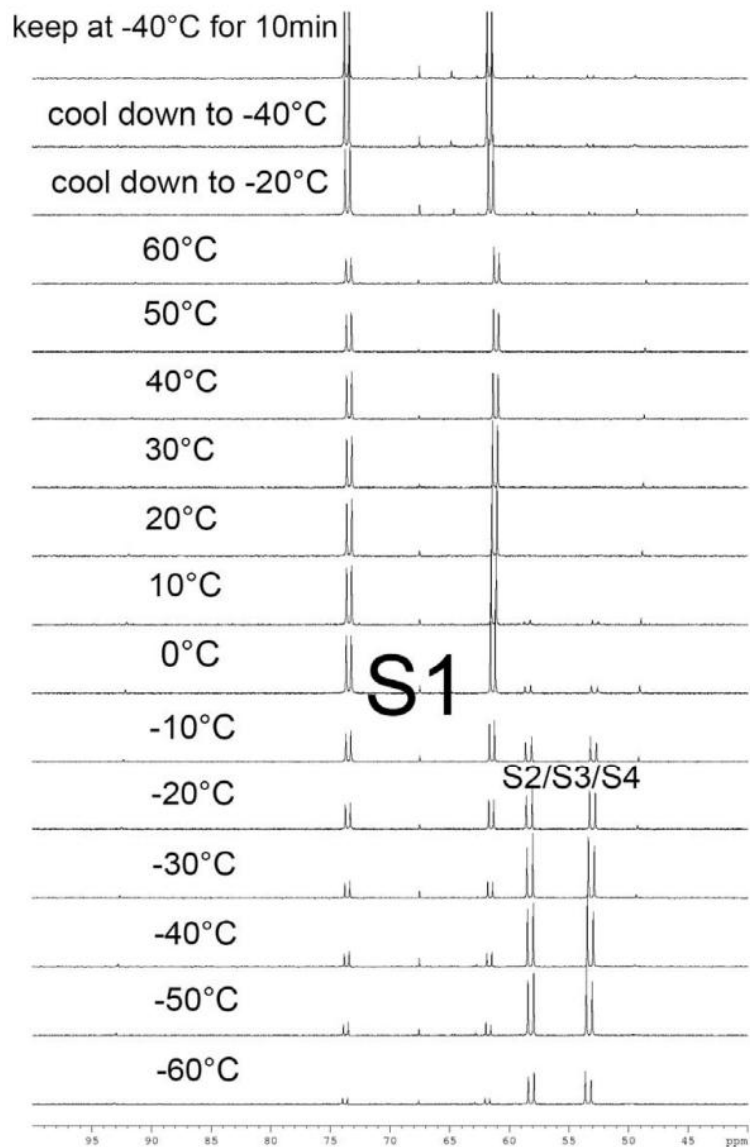
Characterization of Ni Insertion into Benzonitrile



¹H NMR

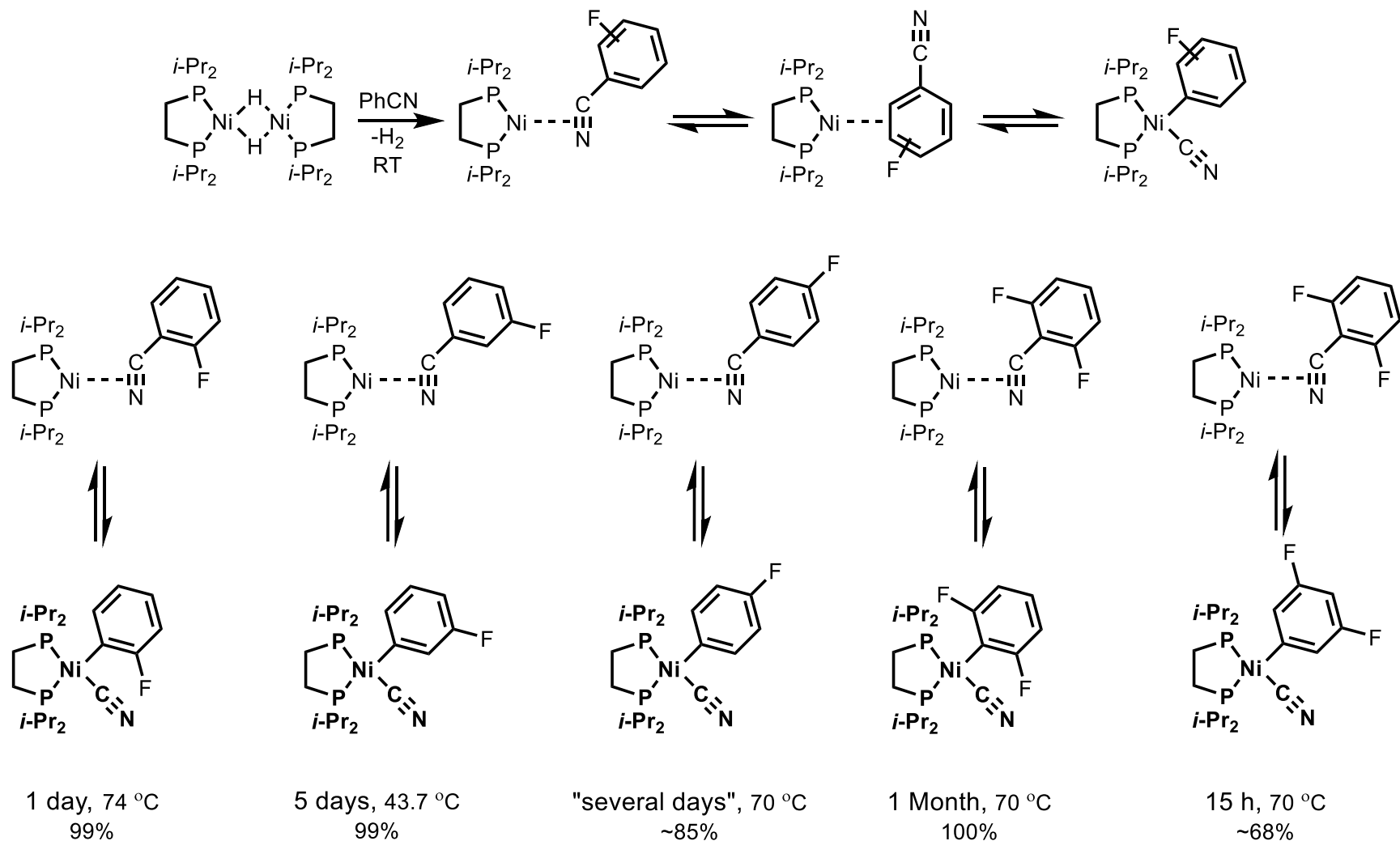


Characterization of Ni Insertion into Benzonitrile

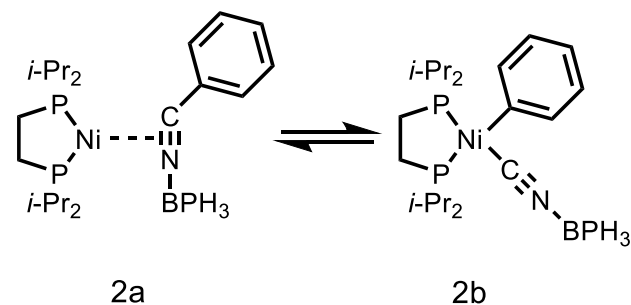
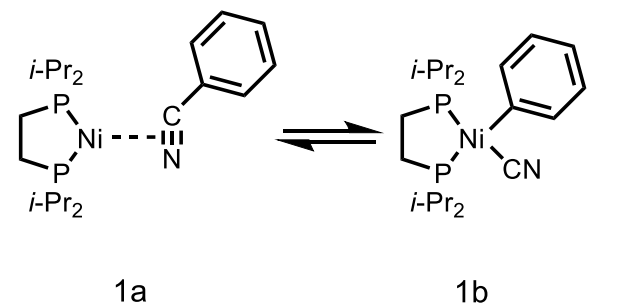
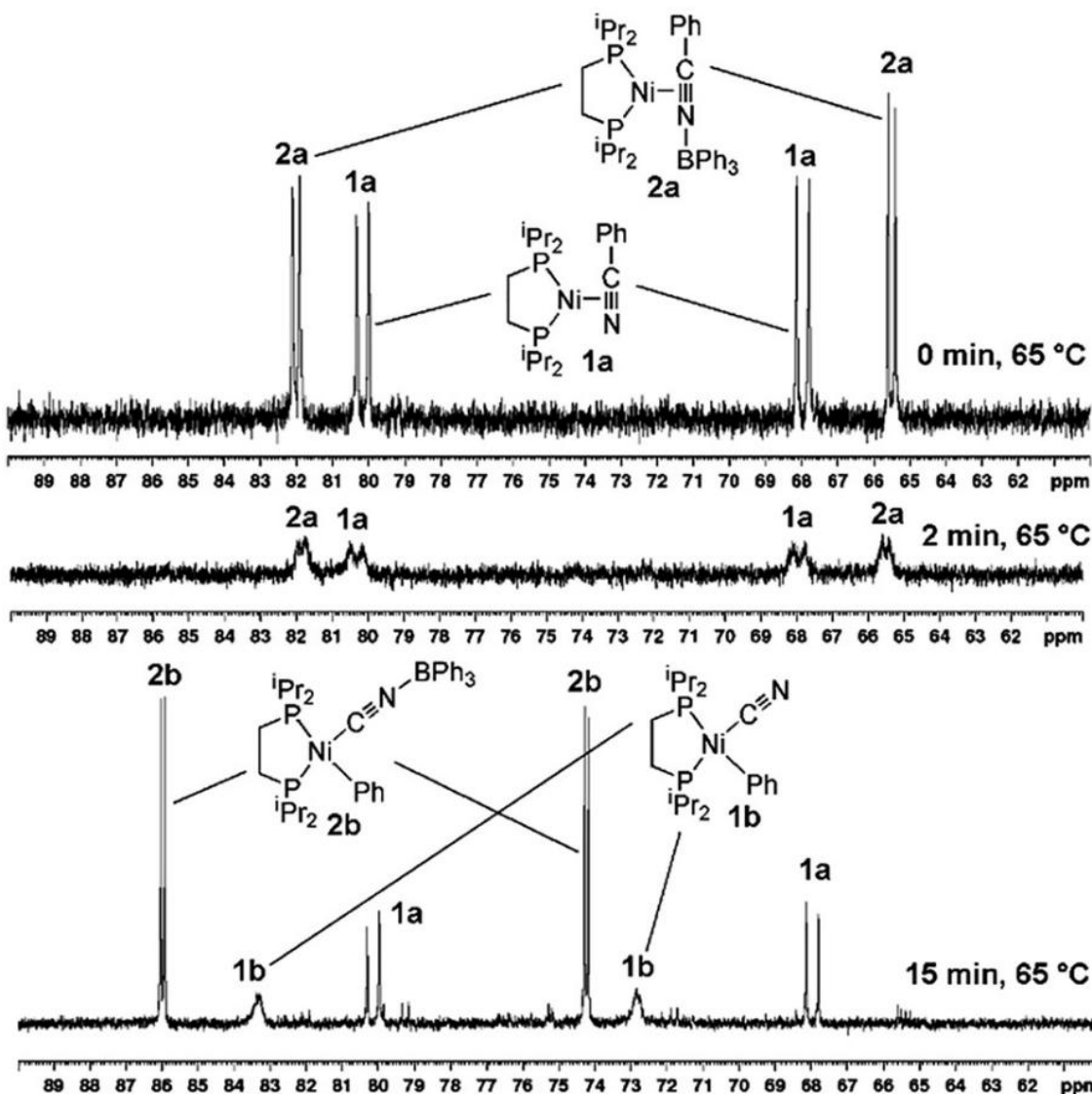


^{31}P NMR

Fluoro-Substituted Benzonitrile C-CN Insertion

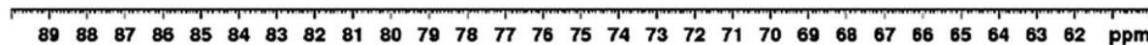
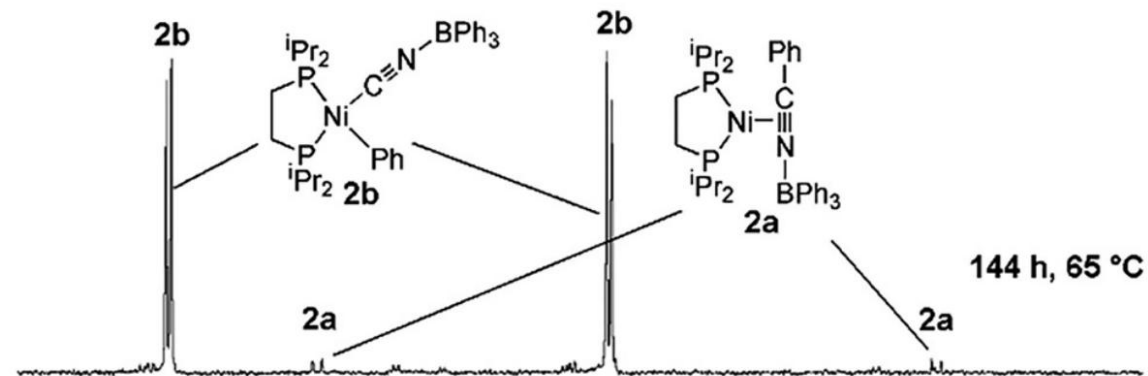
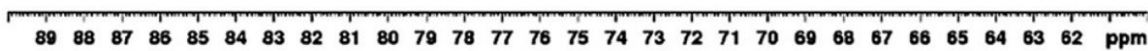
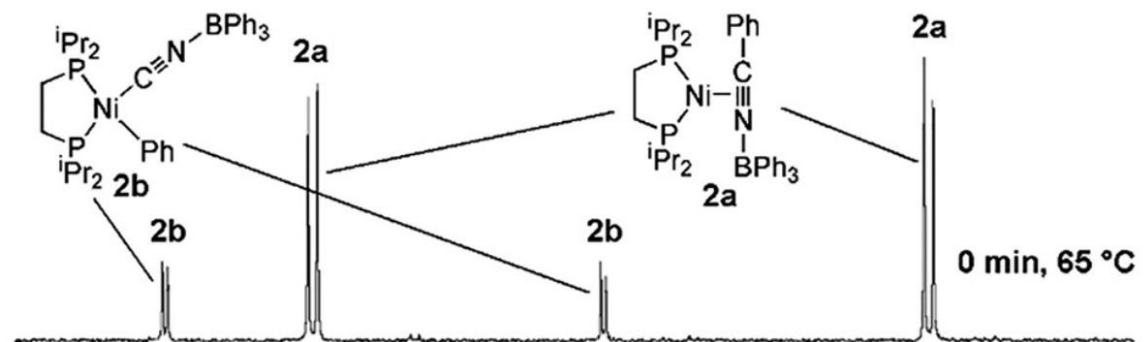


Lewis Acid Assisted Aryl C-CN Cleavage

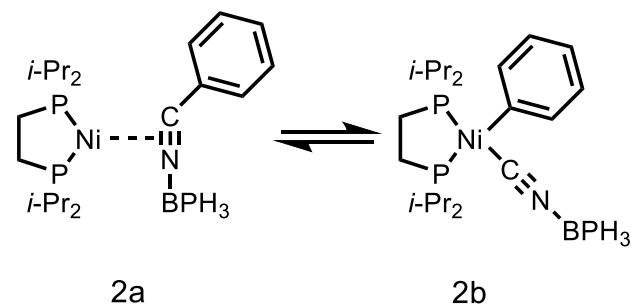
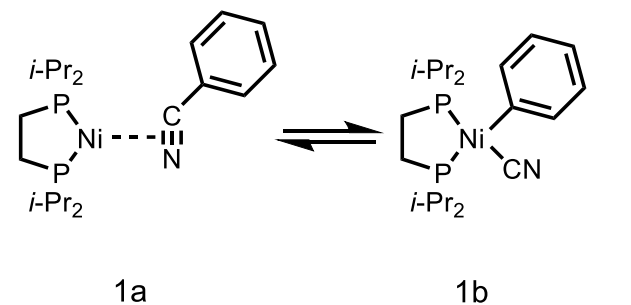


[(dippe)NiH]₂, (1.0 equiv) benzonitrile (2.0 equiv), and BPh₃ (0.5 equiv)

Lewis Acid Assisted Aryl C-CN Cleavage



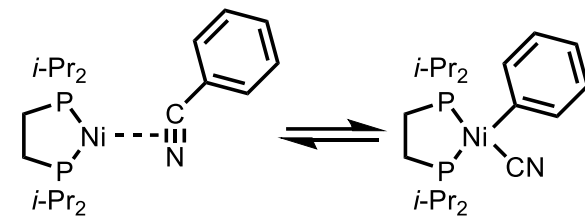
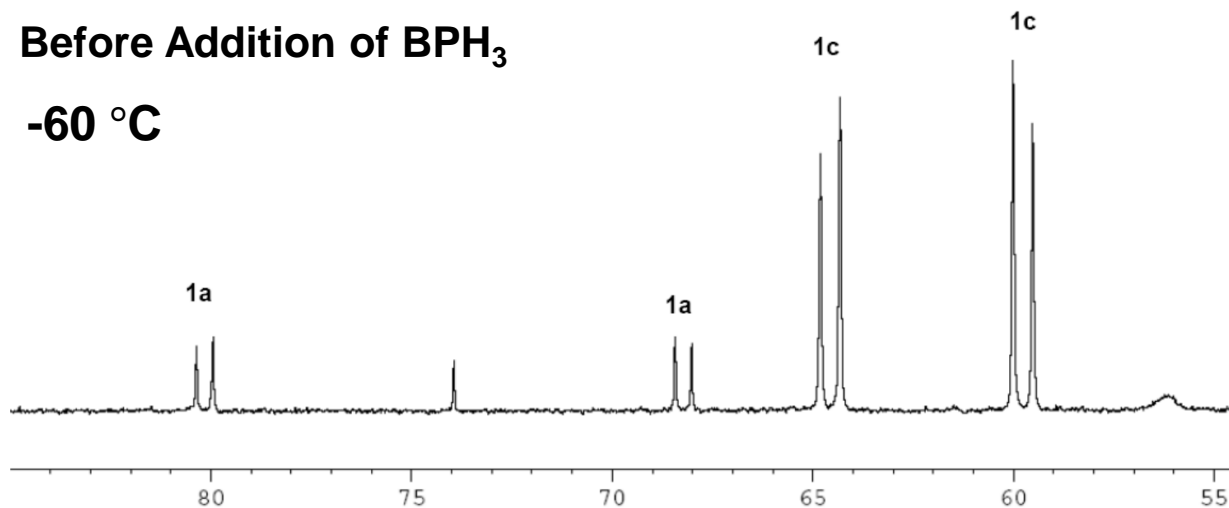
[(dippe)NiH]₂, (1.0 equiv) benzonitrile (2.0 equiv), and BPh₃ (1.0 equiv)



Lewis Acid Assisted Aryl C-CN Cleavage

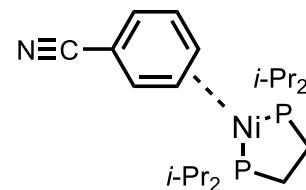
Before Addition of BPH₃

-60 °C



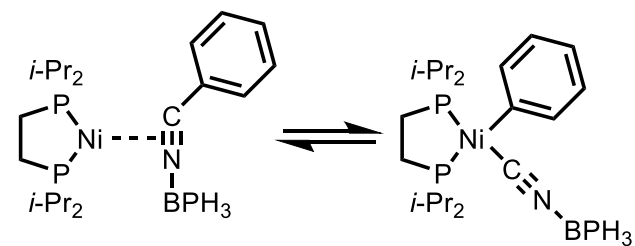
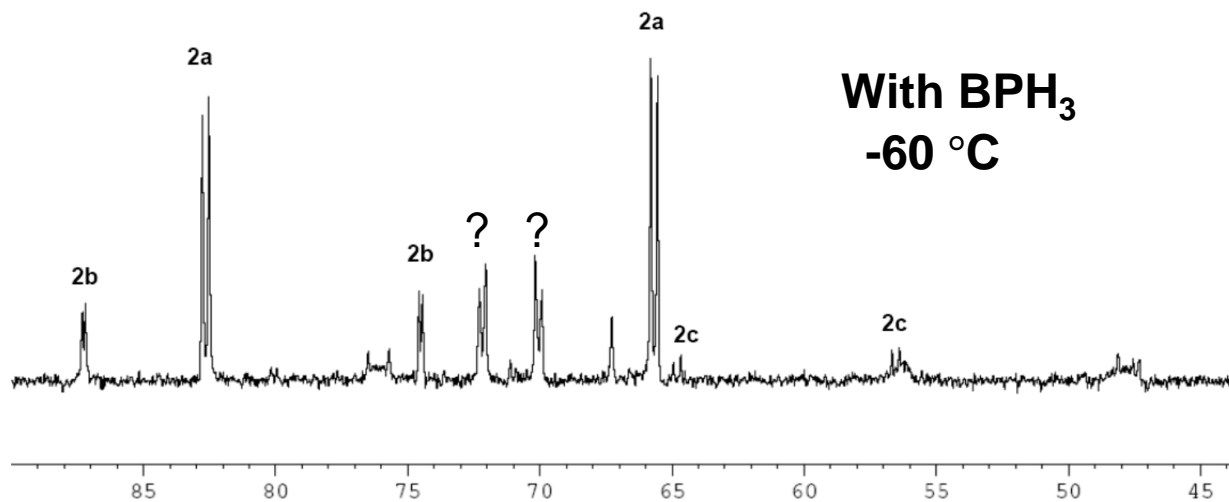
1a

1b



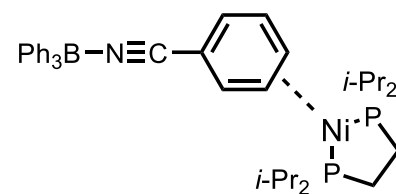
1c

With BPH₃
-60 °C



2a

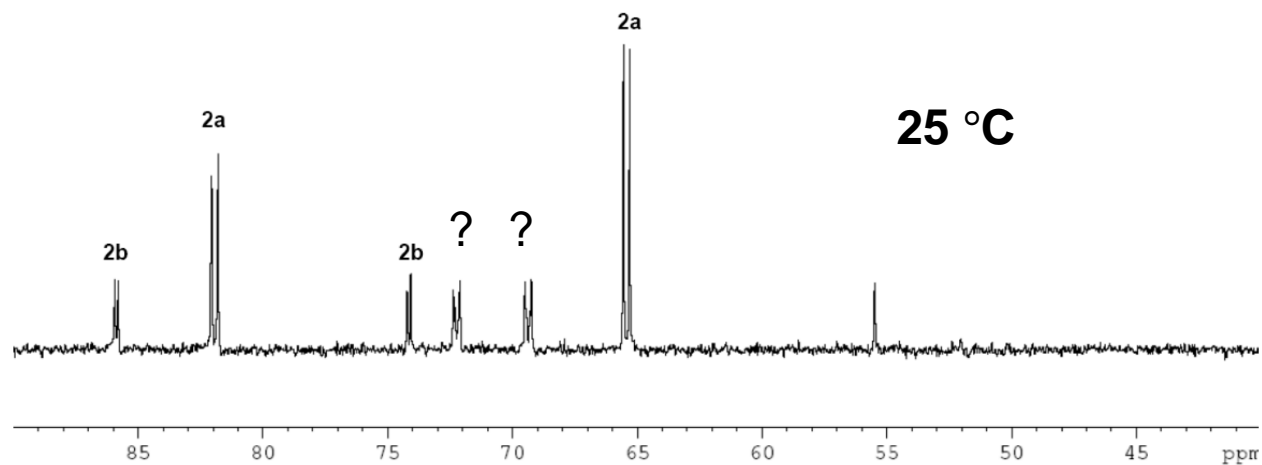
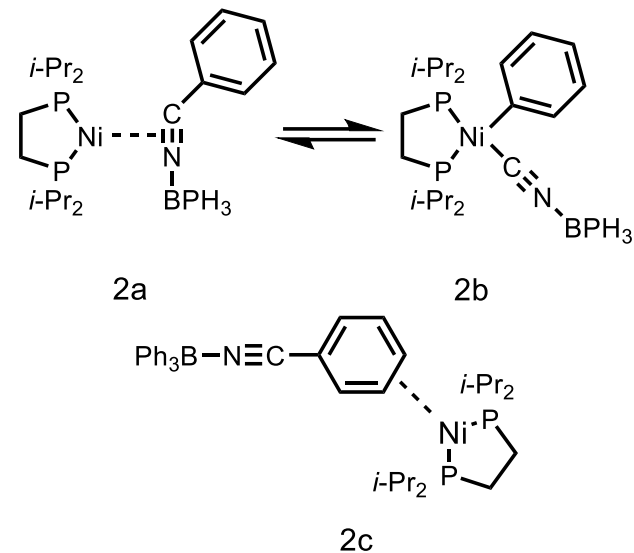
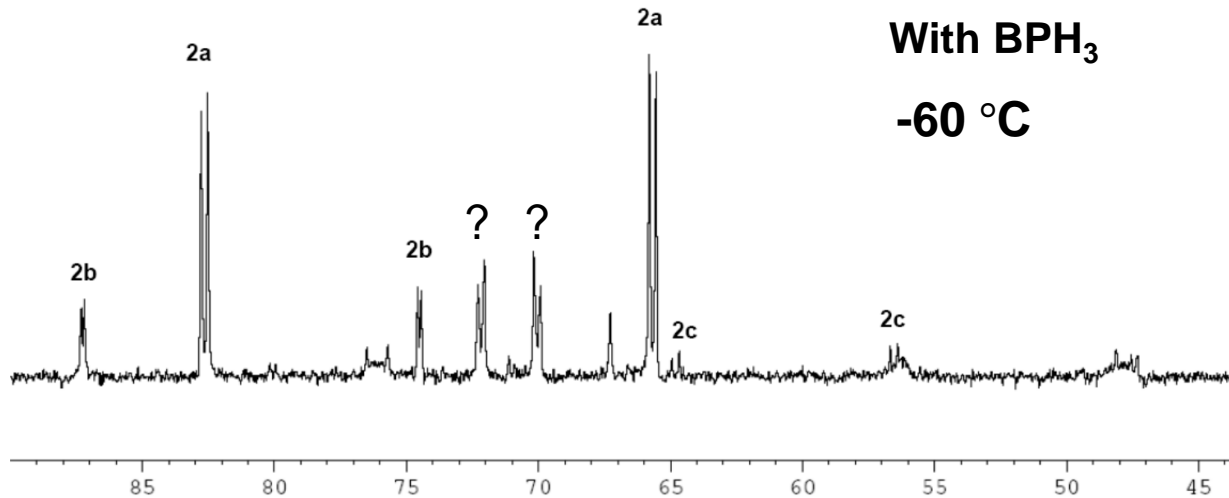
2b



2c

[(dippe)NiH]₂, (1.0 equiv) benzonitrile (2.0 equiv), and BPH₃ (1.0 equiv)

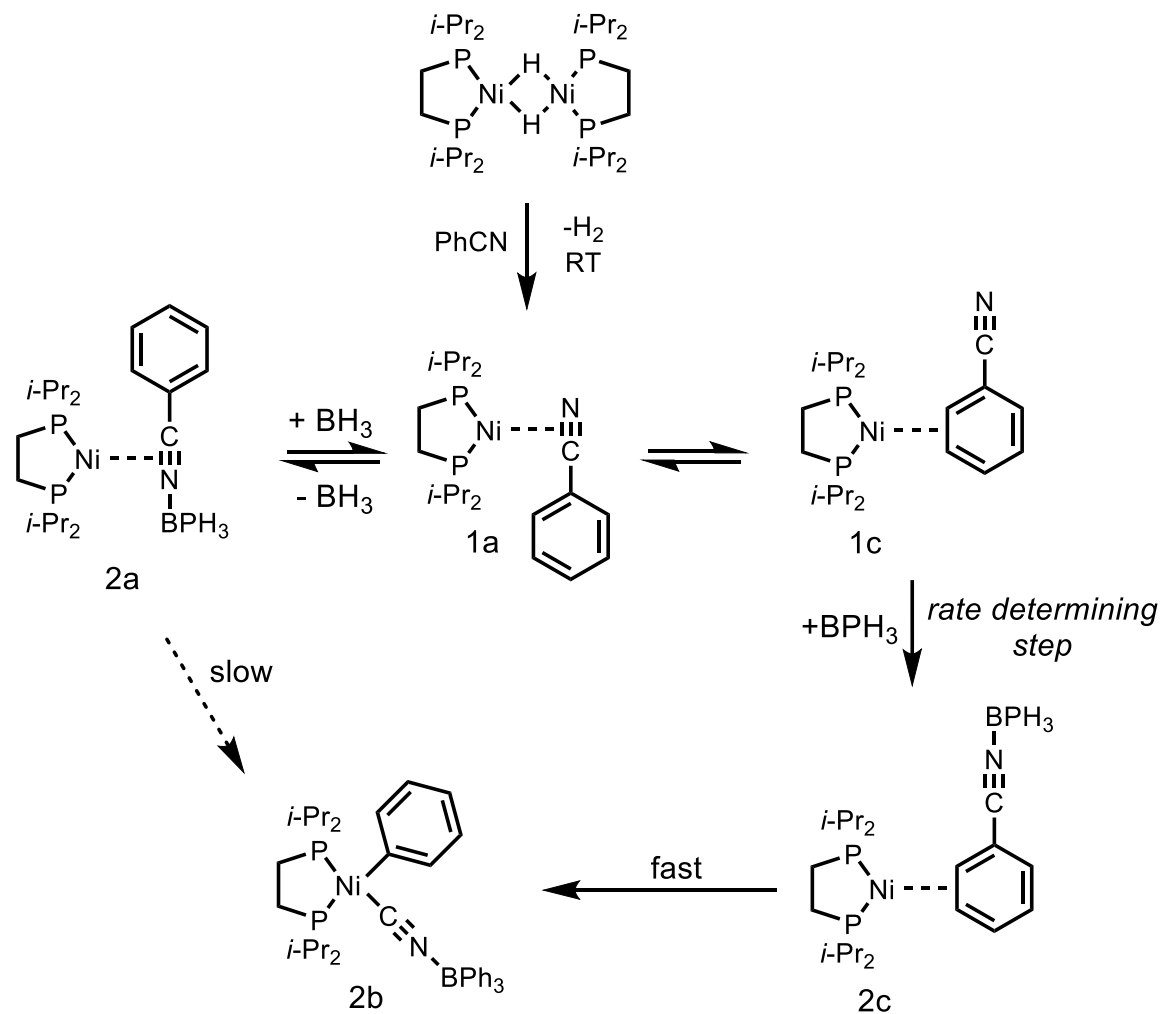
Lewis Acid Assisted Aryl C-CN Cleavage



- 2b, 2c existing concurrently is significant

[(dippe)NiH]₂, (1.0 equiv) benzonitrile (2.0 equiv), and BPh₃ (1.0 equiv)

Proposed Mechanism

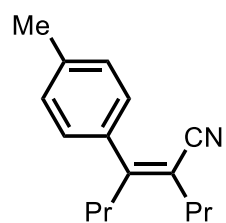
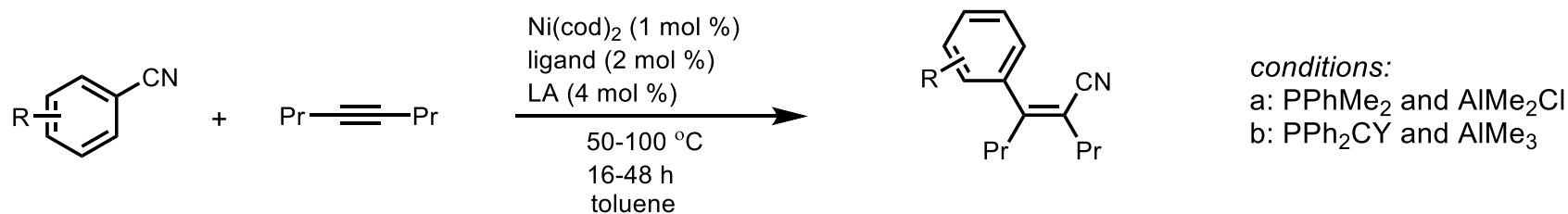


Outline

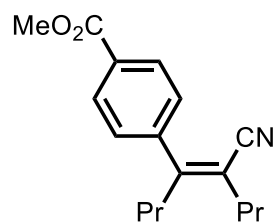
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Part II. Aryl C-CN Bond Activation in Synthesis

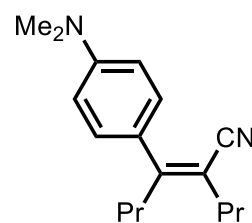
Ni/Lewis Acid-Catalyzed Arylcyanation



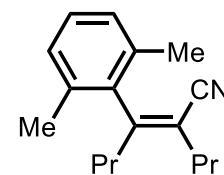
72%^a



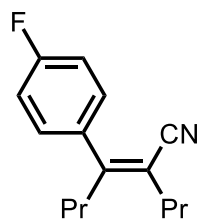
93%^a



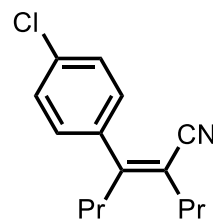
87%^a



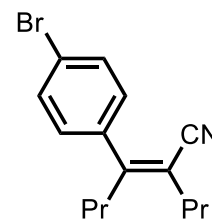
78%^a



95%^b

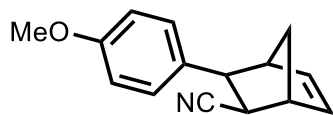
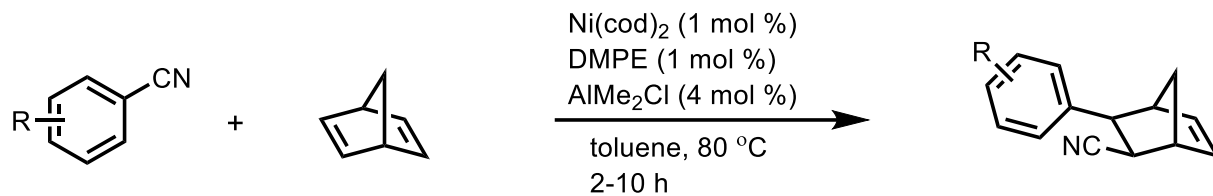


94%^b

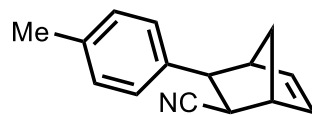


72%^a

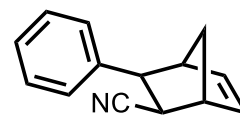
Ni/Lewis Acid-Catalyzed Arylcyanation



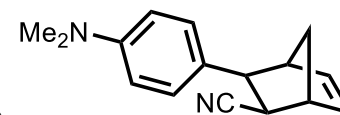
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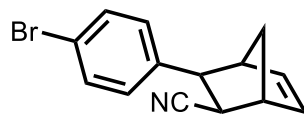
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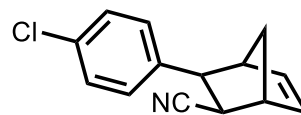
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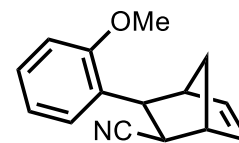
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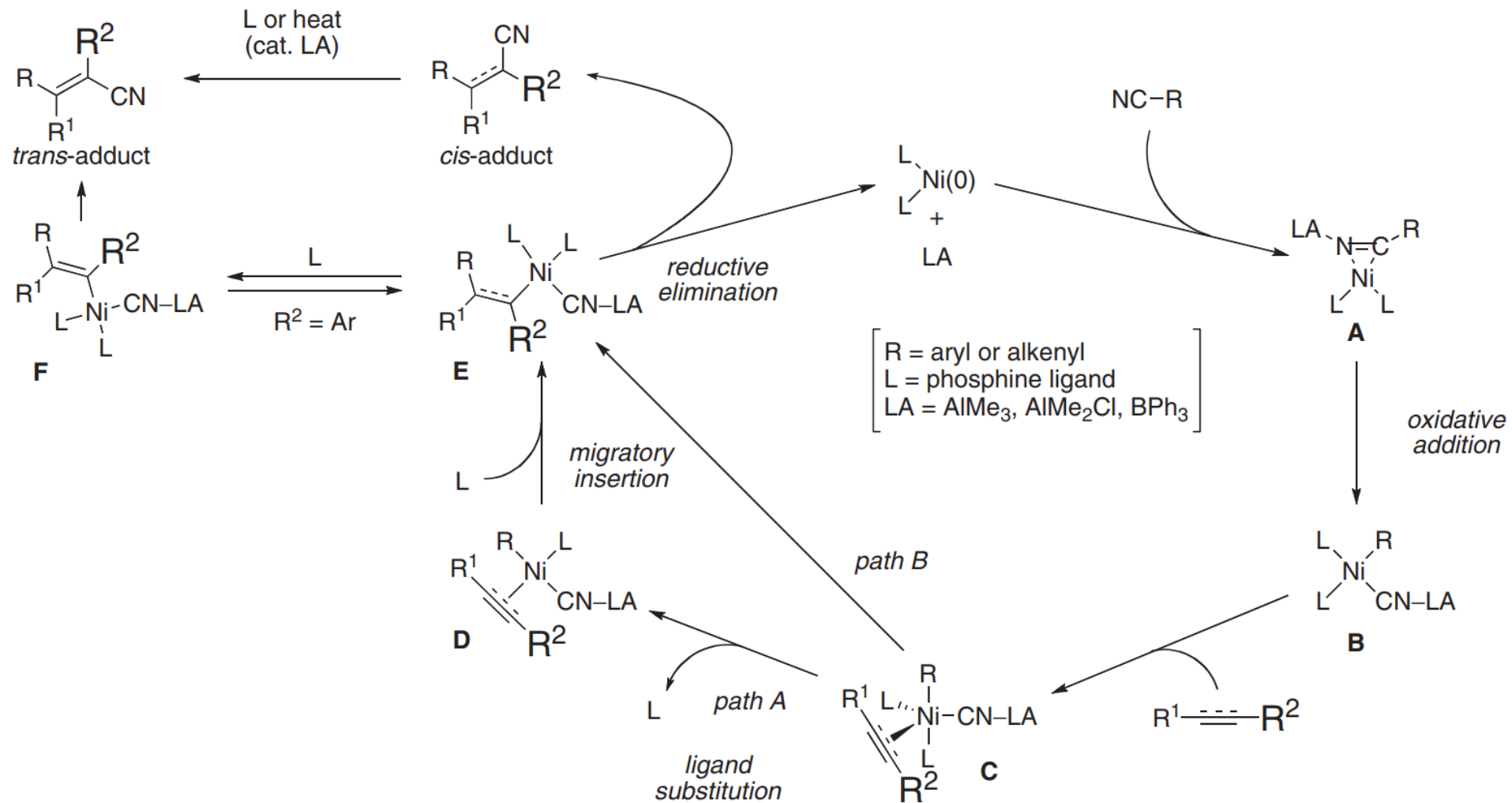


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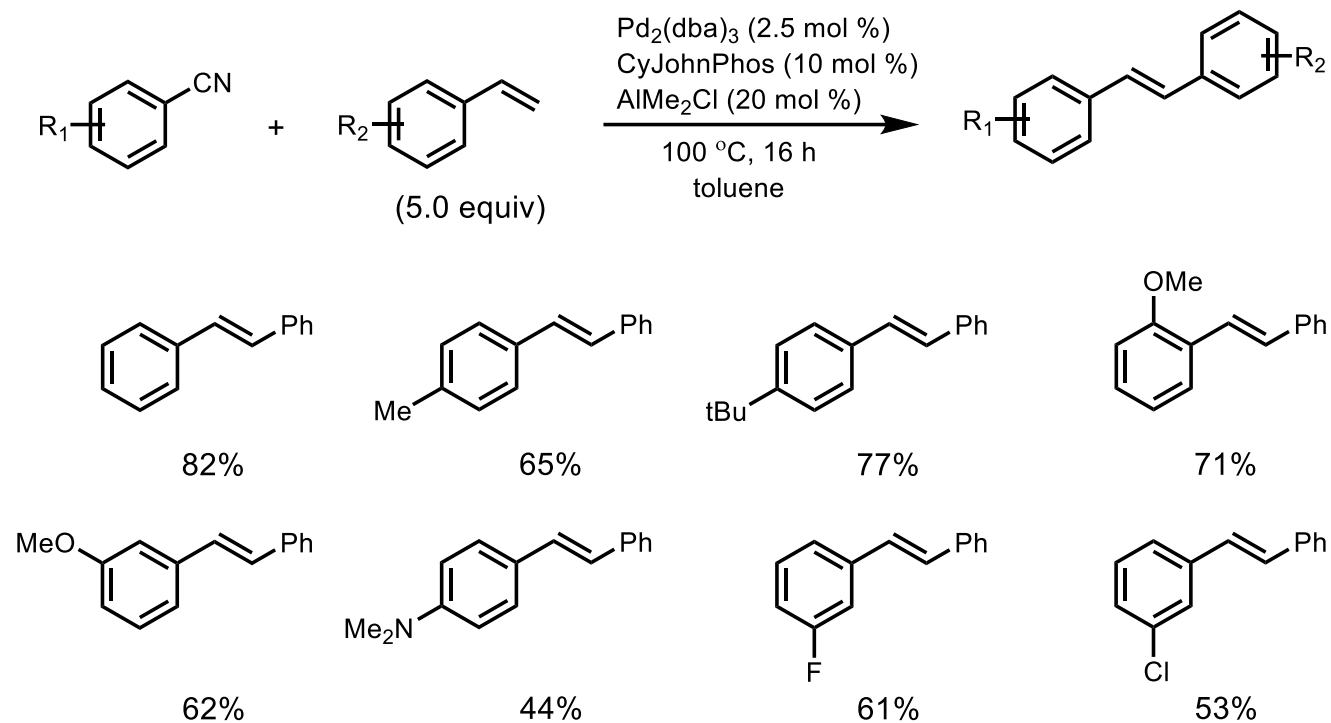


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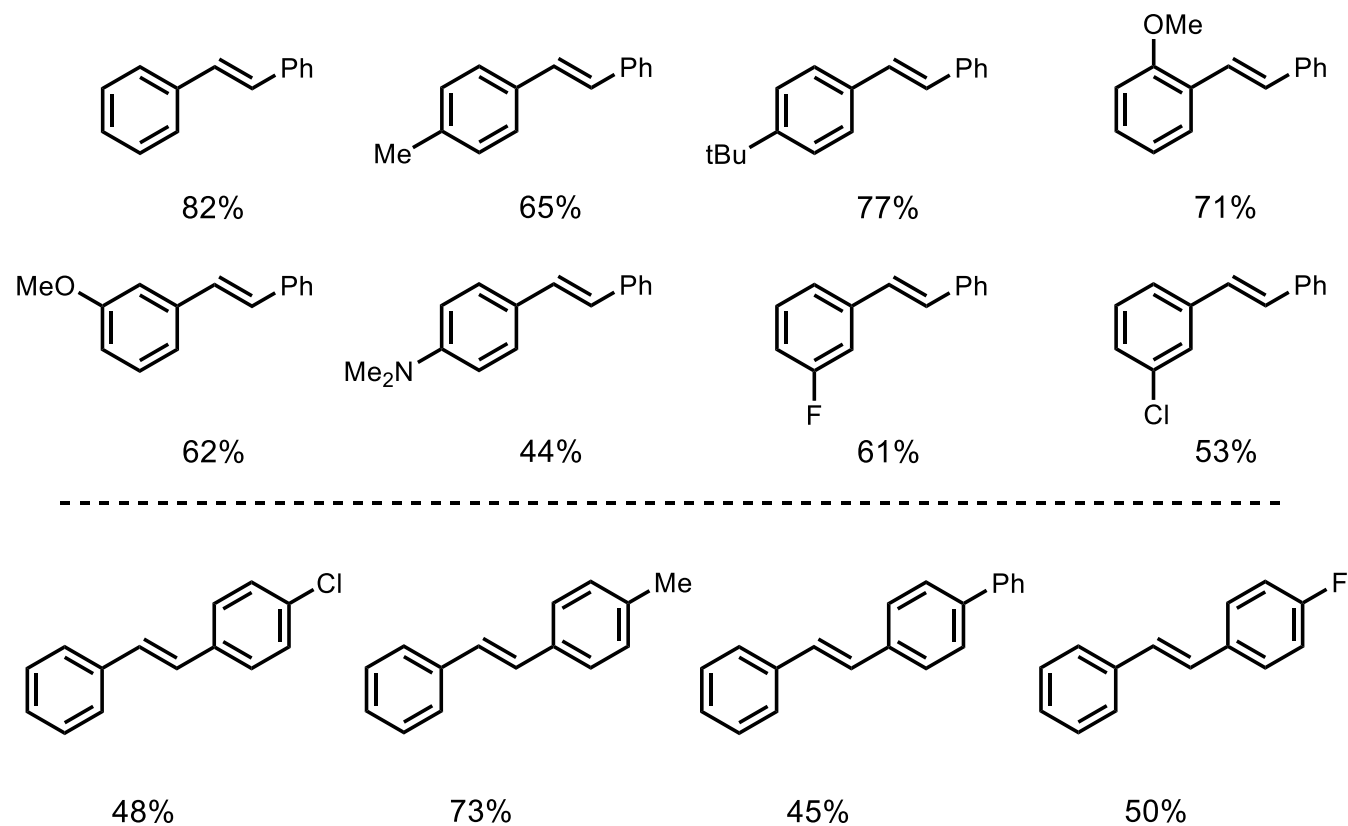
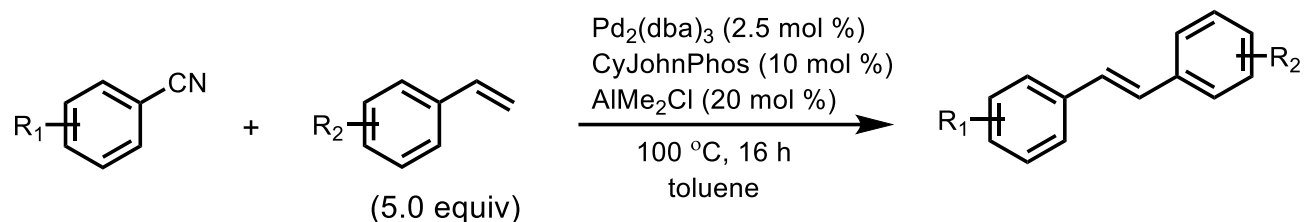
Ni/Lewis Acid-Catalyzed Arylcyanation



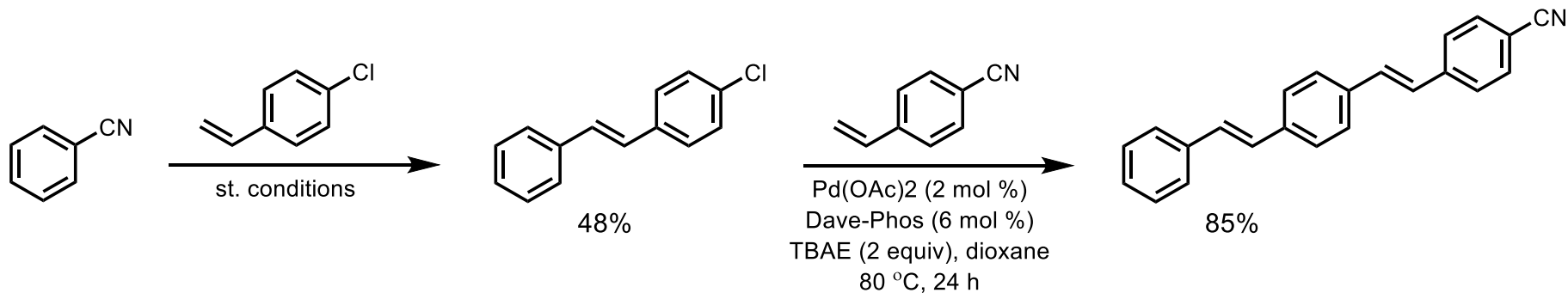
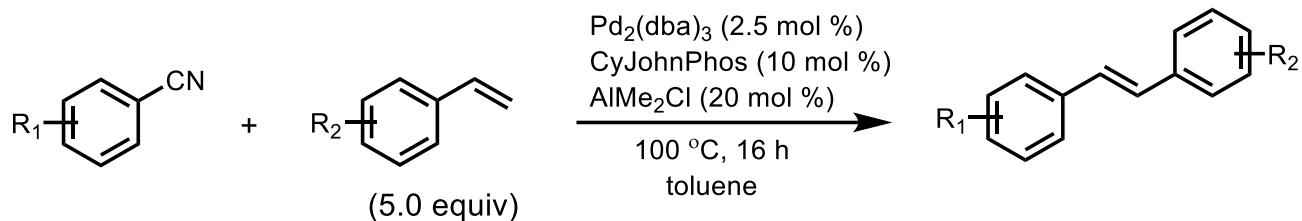
Mizoroki-Heck-Type Reactions of Aryl Cyanides



Mizoroki-Heck-Type Reactions of Aryl Cyanides



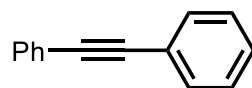
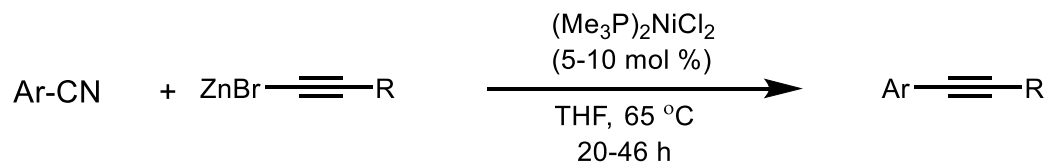
Two-Step Synthesis of Orthogonal Heck Reactions



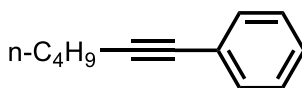
CN cleavage in the presence of Cl

Cl cleavage in the presence of CN

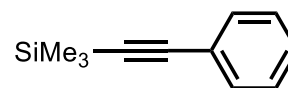
Ni-Catalyzed Alkynylation of Benzonitriles



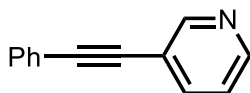
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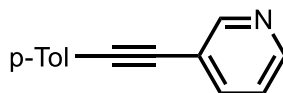
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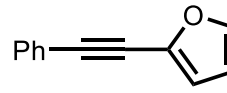
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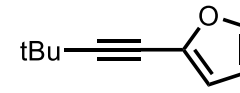
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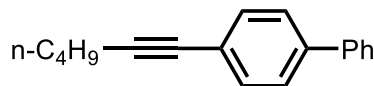
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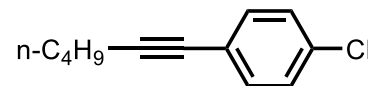
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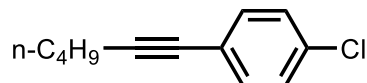
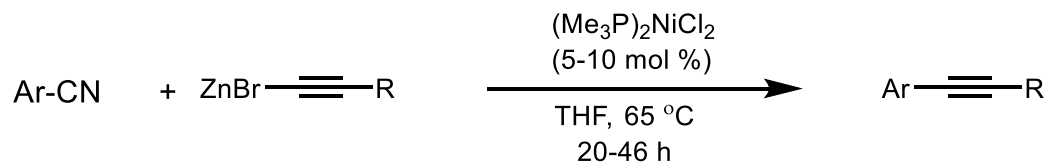


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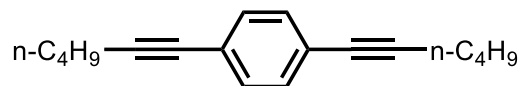


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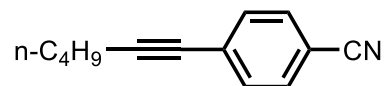
Ni-Catalyzed Alkynylation of Benzonitriles



65%*



5%



N.D.

“Remarkable”

Conclusions

- Low valent Ni(0) systems are very effective at aryl C-CN insertion
 - Lewis Acids greatly accelerate oxidative addition into aryl C-CN bonds
 - Aryl C-CN bonds can be activated, in some cases with excellent chemoselectivity!
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